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Biosynthesis of Silver Nanoparticles from Medicinal Plant, *Pedalium murex* L and Evaluation of its Antibacterial Activity against Selected Pathogens

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Authors' contributions

This work was carried out in collaboration among all authors. All authors read and approved the final manuscript.

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ABSTRACT

Silver nanoparticles play an important role in controlling mosquito population as well as multidrug resistant pathogens. An aqueous extract of *Pedalium murex* L. was used for the synthesis of silver (Ag) nanoparticles and its antibacterial activity. The green synthesis of nanoparticles in plants is cost-effective and eco-friendly approach. To identify the compounds responsible for the reduction of silver ions, the functional groups present in plant extract were investigated by Fourier Transformation Infra Red Spectroscopy (FT-IR) and characterization of nanoparticles was done using standard analytical methods. UV-visible spectrum of the aqueous medium containing silver nanoparticles showed absorption peak at around 350-600 nm. The compounds observed from FT-IR spectra are correlated with the reduction and capping material of silver nanoparticles. The observed results suggested that the bioactive compounds present in *Pedalium murex*L, was found to exhibit a significant inhibitory activity against selected pathogens.

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1. INTRODUCTION

In ancient times, plants have provided a source of idea for the production of new drug and medicines derived from different plants have made to improve the human health and wellbeing [1]. India, a biodiversity nation embraces the indigenous knowledge of traditional herbs. Since decades, the empirical knowledge of utilizing the medicinal plants for the treatment of various ailments was in wide use [2]. The World Health Organization estimated that 80% of people employ the plant extracts as folk medicine in conventional therapies [3]. Furthermore, 900 types of valuable medicinal plants are said to be establish in Nepal among 7,000 species of medicinal plants exist in the world [4,5]. From the different parts of the world, the antimicrobial properties of medicinal plants have been reported [6-8]. The most common infection produced by microorganisms are fever, mouth infections, jaundice, guinea worm sores, joint pain, food borne disease, kidney infection and etc. In recent years, several drugs resistance was developed to human and plant caused by pathogenic microbes due to randomly usage of viable antimicrobial drugs used for the treatment of diseases [9]. These antibiotics had some side effects in the host which includes hypersensitivity reaction, immuno-suppression and some allergic reactions [10,11]. Due to this reason, there is a need to develop different antimicrobial drugs for the treatment of several diseases from a variety of sources, including medicinal plants [12,13].

An extensive research has been carried out recently with secondary plant metabolites [14] for many diseases the treatment of [15]. Phytochemicals are mainly divided into two groups, i.e. Primary and Secondary constituents; based on their functions of plant metabolism. Phytochemicals are natural bioactive compounds present in all medicinal plants. These compound work with nutrients and fibers act as resistance against various diseases and in stress conditions. Primary metabolites contain common sugars, amino acid, proteins and chlorophyll secondary metabolites while consists of Terpenoids, Saponins, Phenolic Alkaloids, compounds, Flavonoids, Tannins etc [16].

The medicinal importance of a plant is due to the presence of bioactive compound such as primary and secondary metabolites and these active compounds usually present in the storage organs of the plants like fruits, roots, seeds, bark, leaves etc. Plant products are always hailed to be a potent remedy against dreadful diseases with lesser or no side effects [17]. Hence, more studies are pertaining to the use of the plant as therapeutic agents should be emphasized; especially those related to the control of antibiotic resistant microbes. Contrary to the synthetic drugs, antimicrobials of plant origin are not associated with many side effects and have an enormous therapeutic potential to heal many infectious diseases.

Nano-biotechnology has emerged as an important division of nanotechnology. One of the important aspects in thefield of nanotechnology is the development of a moreconsistent process for the synthesis of nanomaterials morethan a range of size (with good monodispersed) andchemical composition [18]. Noble metalnanoparticles have been gaining a lot of significance in he past few years due to their applicability in the field ofphysics, chemistry, medicine, biology and material science [19]. Metal nanoparticles have a high specific surface area and surface atoms, because of their physicochemical characteristics, outstanding including optical. catalytic, electronic, magneticand antibacterial properties. Synthesis of metal nanoparticles is enormous due to their potential applicability indifferent areas such as electronics, chemistry, energy, and medicine development [20]. Metal nanoparticles, especially with inert metals, have been extensively studied, due to their strong optical absorption in the visible region, caused by the group excitation of the free electron gas [21].

The silver nanoparticles have nonlinear optics, spectrally selective coating for solar energy absorption, biolabeling, intercalation materials for electrical batteries as optical receptors, catalyst in chemical reactions etc [22]. Silver is well known for possessing an inhibitory activity towards many bacterial strains and microorganisms process [23]. The of synthesizing silver nanoparticles by chemical reduction is as colloidal dispersions in water or organic solvents [24]. Silver and its nanoparticles have broad applications, especially in skin ointments. Various green synthetic approach using plant leaf extracts from Alternanthera sessilis [25]. Morindacitr ifolia [26]. Mukia scabrella [27] Iresine herbstii [28] Tribulus terrestris [29]. Azadirachta indica [30] Cycas

circinalis. Ficus amplissima, Commelina Lippianodiflora [31]. benahalensis. Ocimum sanctumand Aloe vera [32] were widely reported. Pedalium murex is a luscious herb originated from different parts of the globe such as India, tropical Africa, Sri Lanka, Mexico and Pakistan. In India, it occurs primarily in the Western and Corommandal coastal area. Pedalium murex Linn is mainly used in the treatment of disorders urinary systems such as gonorrhea. of incontinence of urine, dysuria, etc [33-38]. It act as antibilious agentand is widely used to promote lochial discharge from juice of fruits. It is also used to control white discharge due to excessive body heat in the decoction of leaf. Pedalin dinatin-7-glucuronide. Glycosides present in the seeds of plant showed the mild diuretic activity.

The secondary metabolite, phytosterol is present only in this plant and large amounts of gums and mucilage is also present. It purifies the blood and removes stone in the bladder. According to Unani system of medicine, Phytosterols find significant therapeutic application as diuretic and it enriches blood and useful against various In thepresently investigation, ailmets. the synthesis of silver nanoparticles from Pedalium murex extract and ascertain L. theircharacterizationand also to evaluate the efficacy of antimicrobial activity against several Gram-Positive and Gram-Negative bacteria.

2. MATERIALS AND METHODS

2.1 Collection and Identification of Plant

The plant *Pedalium murex* (L.) was collected from Thanjavur district, Tamil Nadu in the month of September. It was taxonomically identified and authenticated by Rev Dr. S. John Britto SJ, Director, The Rapinat Herbarium and Centre for Molecular systematic, St. Joseph College (Autonomous), Tiruchirappalli, Tamilnadu, India (Fig 1).

2.2 Preparation of Extract

For aqueous extraction, add 50 g of dried leaf powder immersed in 250 ml of distilled water for 24 hrs at room temperature and then boiled for 15 min on slow heat. Then the residue was removed by filtering through four layer of muslin cloth to obtain filtrate. Again, the filtrate was centrifuged at 5,000 rpm for 15 min and supernatant was collected. Finally, the extract was preserved at 4°C for further use.



Fig. 1. Collection of *Pedalium murex* (L.)

2.3 Synthesis of Zinc Oxide Nanoparticles

2.3.1 Preparation of 1mM AgNO3 Solutions

One milimolar solution of AgNO3 (0.017 gms) was prepared by dissolving in 100ml deionized water (DIW) and stored in amber colored bottle in cool and dry place.

2.3.2 Green synthesis of AgNPs

10 ml of aqueous leaf extract was added into 20ml aqueous solution of 1 mM AgNO3 aqueous solution into 170 ml of DIW for reduction into Ag+ ions at room temperature.

Reduction of Ag+ to AgO was confirmed by the colour change of solution from colourless to brown. Its formation was also confirmed by using UV–Visible spectroscopy.

2.3.3 Preliminary phytochemical screening for plant extract

The presence of flavonoids, alkaloids, glycosides, steroids, phenols, saponins, terpenoid, cardiac glycosides and tannins in the plant *Acalypha indica* (L.) were analysed by the standard methods of [39].

2.4 Alkaloids

2.4.1 Hager's test

About 2 ml of Hager's reagent (Saturated aqueous solution of picric acid) was added to the

extract, and the presence of alkaloids was indicated by the formation of yellow precipitate.

2.4.2 Wagner's test

To the plant extract, few drops of Wagner's reagent was added by the side of the test tube. A reddish brown precipitate confirmed positive test.

2.4.3 Test for Steroids

2.4.3.1 Salkowski's test

About 100mg of Pedalium murex dried extract was dissolved in 2ml of chloroform. Sulphuric acid was carefully added along the sides to form a lower layer. A reddish brown colour ring formed at the interface was an indicative of the presence of steroid.

2.4.4 Test for Cardiac Glycosides

2.4.4.1 Keller killiani's test

About 1mL of glacial acetic acid was added to 100 mg of the plant extract containing one drop of ferric chloride solution. To which, 1mL of concentrated sulphuric acid was added slowly. A brown ring at the interface indicated the presence of a deoxy sugar, which is characteristic of cardenolides.

2.4.5 Test for Saponins

2.4.5.1 Foam test

The extract diluted with 20ml of distilled water was agitated for 15 minutes. Foam formation indicates the presence of saponins.

2.4.6 Test for resins

To 20 ml of plant extract, 5 to 10ml of acetic anhydrite was added and dissolved by gentle heating. After cooling, 0.5ml of H_2SO_4 was added. A bright purple colour confirms the presence of resins.

2.4.7 Test for Phenols

2.4.7.1 Lead acetate test

The extract (50 mg) dissolved in distilled water with 3ml of 10% Lead acetate solution was added. A bulky white precipitate showed the presence of phenol compound.

2.4.8 Test for flavonoids

2.4.8.1 Alkaline reagent test

To the 2 ml of aqueous solution of plant extract add few drops of Ammonium hydroxide. Formation of a yellow fluorescence indicated the presence of flavonoids.

2.4.9 Test for Tannins

2.4.9.1 Lead acetate test

To 5ml of an aqueous extract, a few drops of 1% solution of lead acetate were added. Formation of a yellow or white precipitate revealed the presence of tannins.

2.4.10 Test for Terpenoid

2 ml of chloroform and 1ml of conc. H_2SO_4 was added to 1mg of extract and observed for reddish brown colour that indicated the presence of terpenoids.

2.4.10.1 Fehling's test

Equal volume of Fehling solution A (copper sulphate in distilled water) and Fehling solution B (potassium tartarate and sodium hydroxide in distilled water) reagents were mixed with the plant extract and boiled. The appearance of the brick red precipitate confirms the presence of reducing sugar.

2.4.11 Test for Gum and Mucilage

The extract (100 mg) was dissolved in 10ml of distilled water, then add 25ml of absolute alcohol with constant stirring. The presence of gum and mucilage was confirmed by the formation of white/cloudy precipitate

2.4.12 Test for Quinone

A few drops of conc. H_2SO_4 or NaOH solution was added to the test solution. The change of colour was taken as an evidence for the presence of quinone.

2.4.13 Test for phlobatannins

The test solution was mixed with 1 ml of 1% magnesium acetate solution. A purple colouration was formed to show positive result for the presence of anthraquinone.

2.4.14 Test for Coumarins

One percent of alkaline KOH solution was added to the test solution. The formation of golden yellow colour indicates the presence of coumarins.

2.5 Characterization of Phytochemicals

2.5.1 UV-VIS Spectral analysis

The ZnO-NPs were characterized using various spectroscopic and microscopic techniques. UV–visible spectrum was evaluated using UV–Visible spectrophotometer (Shimadzu UV-2450) and the spectrum was recorded between 300 and 800 nm (Huzaifa *et al.*, 2019).

2.5.2 FTIR Analysis

Fourier transform infrared (FTIR) analysis of the NPs was carried out with Fourier transform spectrometer (Shimadzu FT-IR Prestige-21 Model) at a frequency range of 400–500 cm⁻¹ (Huzaifa *et al.*, 2019).

2.5.3 SEM Analysis

Morphological analysis of the synthesized ZnO NPs coated with platinum was carried out using scanning electron microscope (SEM) (JOEL JSM 6335-F) equipped with 150 kV acceleration voltage, and energy-dispersive (Huzaifa *et al.*, 2019).

2.6 Antibacterial Activity

2.6.1 Test Organisms

For this study, both Gram-Positive [*Staphylococcus haemolyticus* (MTCC 3383) and *Staphylococcus hominis*(MTCC 96s)] and Gram-Negative [*Escherichia coli* (MTCC 433), Vibrio cholerae (MTCC 3906), *Salmonella typhi* (MTCC 733)] bacteria were used to determine the antibacterial activity of different extracts of plant *Pedalium murex*.

2.6.2 Preparing Inoculum

About 1.3 g of nutrient broth (NB) was dissolved in 100 ml of distilled water to prepare the bacterial broth. The bacteria was inoculated in the broth medium and incubated for 18- 24 h at 37°C.

2.7 Determination of Antibacterial Assay

2.7.1 Agar-well diffusion method

Antibacterial activity of plant Pedalium murex extracts was carried out by a modified agar method [40]. Nutrient agar (NA) plates were swabbed (sterile cotton swabs) with Twenty four hour old broth culture of respective bacteria. consequently, using sterile borer, well of 0.5 cm diameter was made into the each agar plate and then 90 µl containing 500 µg/ml concentration of each extract (Aqueous) in aseptic condition filled into the well. The plates were placed at room temperature for an hour to allow diffusion of extract into the agar. Then the plates were incubated for 24 hrs at 37°C. The results were recorded by measuring the diameter of inhibitory zone using a transparent meter rule at the end of the 24-48 hrs. Tetracycline was used as standard drugs for antibacterial activities, respectively.

3. RESULTS AND DISCUSSION

Phytochemical evaluation of the silver nanoparticles of aqueous leaf extract of Pedalium murex *L*.were done for the presence of steroids, alkaloids, phenols, tannins, flavonoids, Terpenes, Cardiac glycosides, saponins, Gum and mucilage and absence in the Carbohydrates, Phlobatannins and results were indicated in Fig. 2 &Table 1.

The active component alkaloids contain \propto tocopherol and Isatin which act as antioxidant and antifungal properties. The bioactive compound, Saponin antifungal. possess anticancer, anti-inflammatory and antifungal properties. Tannin also possesses secondary metabolites which contains antimicrobial and antioxidant properties [41]. Flavonoids contain Dinatin-7-glucuronide, Diosmetin-7alucuronide, Pedalitin [42] which act as antimicrobial, anti allergic and anti-cancer properties. The primary antioxidants or free radical scavenger's properties are present mainly in the metabolites such as tannins and flavonoids which are major groups of compounds [43].

Steroid which contain diosgenin component play a vital role in antimicrobial activities. Anthraquinones also has antimicrobial properties [44] antidepressant, antiparasitic [45] and bacteriostatic. The medicinal plants with gum and mucilage are widely used as pharmaceutical adjuvant, due to its properties to act as a suspending agent [46] These secondary metabolites play a vital role in the antimicrobial activity against human pathogenic bacteria which is used in these studies.

The reduction of AgNO3 was visually evident from the color change (Brownish-Yellow) of the reaction mixture after 24 h (Fig. 3). The intensity of the brown color increased in direct proportion to the incubation period. This may be due to the excitation of the Surface Plasmon Resonance (SPR) effect and the reduction of AqNO3 [47] The control AqNO3 solution (without Pedalium murex extract) showed no color change. The silver nanoparticles obtained were characterized UV-visible spectroscopy by and the characteristic absorption peaks at 350-600 nm in the Spectrum confirmed the formation of silver nanoparticles. In the present study After 24 h incubation in dark room condition, the light colored reaction mixtures turned into dark brown for indicating AgNPs formation. The surface plasmon resonance (SPR) of AgNPs produced a peak at 440 nm, which suggests the dispersal of AaNPs. which corroborates with the characteristic peaks of silver nanoparticles [48].

The phytochemical analysis of Pedalium murex indicates the presence of flavonoids, alkaloids, steroids, rosins, saponins and proteins. Fig. 2 reveals that strong absorbance were observed at peak 3379.29cm⁻¹, 3163.26 cm⁻¹, 3005.10 cm⁻¹, 2627.05 cm⁻¹, 2252.86 cm⁻¹, 2148.70 cm⁻¹, 1442.75 cm⁻¹, 1375.25 cm⁻¹, 1037.70 cm⁻¹, 918.12 cm⁻¹, 748.38 cm⁻¹ which are are characteristic of N-H stretch, $-C \equiv C-H$: C-H stretch, =C-H stretch, H-C=O: C-H stretch, C≡ N stretch, $-C \equiv C$ - stretch, C-H bend, C-H rock, C-N stretch, O-H bend, C-Cl stretch and Presence of functional groups such as 1°& 2° amines, amides, alkynes (terminal), Alkenes, Aldehydes, Nitriles, Alkanes, aliphatic amines, carboxylic acids and alkyl halides. In overall spectral profile is similar except for the variation in intensities of the bands. The most widely used modes in protein structural studies are 1°& 2° amines and amides. The broad band at 3297 cm⁻¹ has been assigned in the present study to O-H stretching, the bands of proteins have made a small contribution to it. The bands observed at 2923 cm^{-1} and at 3250 cm^{-1} confirms the presence of asymmetric and symmetric starching modes of the methylene chain in the membrane lipids [46]. The sharp bands observed at 1653 cm⁻¹ are assigned to the in plane C=O stretching vibration (amide) and to the aliphatic amines and carboxylic acid [47]. The phenolicgroups participating in ion replacement response are placed in the 1315–1037 and 1456–1600 cm-1 regions for the plant extract [48].

The very strong band at1592 cm-1 is due to C=C stretching in the aromatic ring, confirming the presence of the aromatic group [44]. The silver nanoparticles of O-H stretching in carboxylic acids vibration is shifted from 3785 to 3881 cm-1. The immediate reduction and capping of silver ion into silver nanoparticles in the present analysis might be due toflavonoids and proteins. The flavonoids present in the leaf extract act as a strong reducing agents, which may be suggestive of the formation of AgNPs by reduction of silver nitrate. The flavonoid compounds in the water extract of *M.pendans*might be actively involved and responsible for thereduction of Agto Ag0. The involvement of water-soluble flavonoid in the reduction of metalions using plant extracts is also evidenced from anotherstudy [45].

In SEM analysis AgNPs have been observed with quasi-spheres shape with uniform distribution (Fig. 5). Nevertheless, agglomeration of the particles was observed probably due to the presence of a weak capping agent, which stabilizes the nanoparticles. This may be due to availability of different quantity and nature of capping agents present in the AaNPs synthesized pedalium murex [48-55]. The above findings were corroborative by the observations from the peaks obtained in the FTIR analysis.

The antibacterial property of silver nanoparticles and aqueous extract of whole plant *Pedalium murex* was analysed against bacterial pathogens using tetracycline as control [56-61]. Out of these five bacterial pathogens four were found to be Gram-Negative (*Escherichia coli, Vibrio cholera, Salmonella*) and one were Gram-Positive (*Staphylococcus haemolyticus, Staphylococcus hominis*). Agar well method was used to evaluate the antibacterial activity of taken samples. After twenty-four hours the minimum inhibitory zone of aqueous extract and silver nanoparticles of P. *murex and* control were measured (Table 3& Fig. 6).

Escherichia coli were tested in *P. murex* were found to be AgNPs of aqueous extract with a maximum inhibitory zone (15 mm each), followed by aqueous extract did not show any inhibition against of *E. coli* and control shows of maximum inhibitory zone (33.3 mm each). *Vibrio cholerae*was found to be more susceptible towards the AgNPs of aqueous extract of *P. murex* maximum inhibitory zone (15 mm), followed by control (42.3 mm), and aqueous extract did not show any inhibitory against in *Vibrio cholerae*. *Staphylococcus haemolyticus* were tested were found to be that AgNPs has highest significatory effects (18.3 mm) and aqueous extract of *P. murex* did not show any inhibitory zone and control shows maximum zone (38.3 mm each).

Salmonella typhi was found to be more susceptible towards the AgNPs of aqueous extract of *P. murex* has some inhibitory zone (15 mm), followed by control (40 mm), and aqueous extract did not show any inhibitory against in Salmonella typhi (Fig. 4). Staphylococcus hominiswere found to be AgNPs of aqueous extract with a maximum inhibitory zone (15mm each), followed by aqueous extract did not show any inhibition activity and control shows of maximum inhibitory zone (43.3 mm each).

The well diffusion method maximum susceptibility with 15-20 mm zone of inhibition was observed at the level of 500 µg/ml concentrations against tested bacteria. It was clearly evident that our experimental plant has the antimicrobial/antibacterial property against bacterial species. Likewise, in-vitro antimicrobial activity of methanolic extract of showed 17 mm, 12 mm and 17 mm inhibitory zone of diameter against S. aureus, E. coli and K. pneumonia respectively and there was no zone of inhibition observed in aqueous extract [62-69]. Likewise, methanolic extract Solanum the of

palinacanthum revealed the zone of inhibition against A. hydrophila, B. subtilis and S. aureus. In terms of specific inhibition by petroleum ether extract of Capparis zeylanica produced the inhibitory zone ranging from 10 to 16 mm at a concentration of 16.5 μ g/ml against S. aureus, B. subtilis, K. pneumoniae and P. vulgaris, whereas chloroform, ethanol and water extracts showed inhibitory activity against all six bacterial strains at concentrations of 13.5, 14.0 and 14.0 μ g/ml respectively. MIC value also revealed that almost all tested bacterial strains were sensitive to the ethyl acetate and petroleum ether extracts of our study plant P. murex.

The results showed that the aqueous plant extract have not possessed any antibacterial activity. These results supported that the organic solvent extract is better than aqueous extracts. Similarly, the ethanolic extract of Punica granatum was most active against E. coli and the methanolic extract of Euphorbia fusiformis root possessed significant antibacterial activitv Е. against Staphylococcus aureus. coli. Pseudomonas aeruginosa,K. pneumonia, Proteus vulgaris, S. typhii A and S. typhii B; Solanum torvum showed antimicrobial activity against B. subtilis, B. cereus, P. aeruginosa and S. aureus, while S. nigrum was active against Salmonella typhi. Now, the present study also supported that the ethyl acetate and petroleum ether effectively controls the bacterial growth than the other extracts. This probably indicated that the bioactive ingredients are able to inhibit the growth of the common pathogens.

Compound	AgNPs sPM
Alkaloids	++
Flavonoids	+
Steroid	+
Saponins	+
Quinones	+
Resins	+
Phenols	+
Cardiac glycosides	+
Tannin	-
Terpenes	+
Phlobatannins	-
Carbohydrates	-
Gum and Mucilage	+
Coumarin	+

Table 1. Preliminar	y phytochemical	analysis for	AgNPs of	Pedalium	murex
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(+) Present; (-) Absent

Sivaranjani et al.; JPRI, 33(64B): 76-91, 2021; Article no.JPRI.82499

-1 Frequency, cm	Bond	Functional group
-1 3390.86 cm	O–H stretch, H–bonded	Alcohols, phenols
⁻¹ 2920.23 cm	C≡N stretch	Nitriles
2850 79 cm	C–H stretch	Alkanes
1456 26 cm	C–H bend	Alkanes
1381 03 cm	C–H rock	Alkanes
-1	C–H wag (–CH2 X)	Alkyl halides
1037 70 cm	C–N stretch	Aliphatic amines
545.85 cm	C–Br stretch	Alkyl halides

 Table 2. FTIR analysis of AgNPs synthesized Pedalium murex

Table 3. Antibacterial assessment of AgNPs of Pedalium murex against bacterial pathogens

S. No	Antibacterial activity of AgNPs of <i>Pedalium murex</i> L. in mg/ml (Mean ± S.D)*						
	Pathogens	AgNPs APM	APM	Tetracycline	DMSO		
1.	E. coli	15.0±0.00	-	33.3±5.77	-		
2.	Vibrio cholerae	15.0±0.00	-	42.3±6.42	-		
3.	Staphylococcus haemolyticus	18.3 ± 2.88	-	38.3±2.88	-		
4.	Salmonella typhi	15.0±0.00	-	40.0±0.00	-		
5.	Staphylococcus hominis	15.0±0.00	-	43.3±11.5	-		



Fig. 2. Preliminary phytochemical analysis of synthesized AgNPs Pedalium murex.

Sivaranjani et al.; JPRI, 33(64B): 76-91, 2021; Article no.JPRI.82499



Fig. 3. UV spectra of AgNPs of Pedalium murex L.



Fig 4: FTIR analysis of AgNPs synthesized Pedalium murex



Fig. 5. SEM Analysis of AgNPs of Pedalium murex AgNPs PM– Silver nanoparticles of Pedalium murex L extract; APM – Aqueous extract ofPedalium murex L.; Tetracycline – Positive control; DMSO –Dimethyl sulphoxide (Negative control); *Three replicates of mean



Fig. 6. Antibacterial activity of AgNPs of aqueous extract of *Pedalium murex E. coli; V. cholera; S. parahaemolyticus; S. typhi; S. hominis*

According to this study, plant based antimicrobial drug with silver nanoparticles have enormous therapeutic potential as they can serve the purpose with lesser side effects that are often associated with synthetic antimicrobial agents. The results revealed that the extract of AgNPs P. murex effectively produced inhibitory activities against both gram positive and gram negative bacteria. Presence of chemical components of P. murex associated with silver nanoparticles may inhibit the bacterial growth. These bacterial group incubations around the wall are due to the release of diffusible inhibitory compounds from silver nanoparticles. These biosynthesized nanoparticles are widely used in cancer therapy, wound healing, antimicrobial activity, water paints, cotton fabrics and textiles, etc. The green synthesis of AqNPs has also paved a better methodological approach in the medical field. The present study provides the scientific information about the plant extract of P. murex and supports the usage of this plant for folkloric treatment of traditional healers. Now, the trend is switched over towards the animal disease management. The results of the present study explore the antibacterial potential of P. murex which leads to further study in this direction.

4. CONCLUSION

The study proves to be an eco-friendly, rapid green approach for the synthesis providing a cost effective and an efficient way for the synthesis of silver nanoparticles. Therefore, this reaction pathway satisfies all the conditions of a 100% green chemical process. The biosynthesized AgNPs were found to have a pronounced antibacterial activity against selected pathogens. In this present study, proteins and flavonoids in the Pedalium murex extract play an important role in the formation of silver nanoparticles. Such eco-friendly method could be a better alternative to the conventional physical/chemical methods used for the synthesis of silver nanoparticles and thus has a potential to use in biomedical applications. Hopefully, such green therapeutic approaches will play an important role in optoelectronics and medical devices in the near future.

CONSENT

It is not applicable.

ETHICAL APPROVAL

It is not applicable.

NOTE

The study highlights the efficacy of "traditional herbs" which is an ancient tradition, used in some parts of India. This ancient concept should be carefully evaluated in the light of modern medical science and can be utilized partially if found suitable

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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